

EXSOLUTION TEXTURES IN HIGH PRESSURE GARNETS, MIR KIMBERLITE, SIBERIA

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One of the more intriguing findings of the last 15 years or so is the recognition of very high pressure rocks and minerals. These occurrences include: ultrahigh-pressure metamorphic terranes containing minerals (e.g., diamond, coesite) indicative of pressures higher than 30 kbar (e.g., Smith, 1984; Sobolev and Shatsky, 1990; Wain, 1997) (2) very high pressure mineral inclusions (e.g., majoritic garnet, ferropericlasite) in diamonds from kimberlites (e.g., Tsai et al., 1979; Moore and Gurney, 1985; Stachel et al., 2000), and (3) exsolved phases in host minerals that suggest that the original mineral had a composition indicative of very high pressures (Haggerty and Sautter, 1990; Zhang and Liou, 1999).

Proposed examples of exsolution due to decompression (#3 above) from very high pressure include rods of ilmenite in olivines from the Alpe Arami garnet peridotite massif (Dobrzhinetskaya et al., 1996) and the Dabie-Sulu ultra high pressure terrane (Hacker et al., 1997), K-feldspar lamellae in clinopyroxene and coesite in titanite from diamond-bearing rocks of the ultrahigh-pressure Kokchetav massif, Kazakhstan (Sobolev and Shatsky, 1990; Ogasawa et al., 2002), K-feldspar lamellae in clinopyroxene inclusions in eclogitic garnets from kimberlites (Harlow and Veblen, 1991) and rods of pyroxene in garnet from kimberlite xenoliths (Haggerty and Sautter, 1990) and orogenic garnet peridotites from the Western Gneiss region of Norway (van Roermund and Drury, 1998).

Some elemental substitutions in garnet are also thought to be indicative of high pressure, and to result in exsolution of various minerals if the garnet reequilibrates at low pressures. Experimental evidence is most clear for the incorporation of supersilicic, majoritic components (MSiAl_2 where $M = \text{Fe}^{2+}$, Mg, Ca) in garnets at pressures greater than 50 kbar (e.g., Akaogi and Akimoto, 1979; Irifune 1987; Gasparik, 1989). A second important mechanism for producing supersilicic garnets at high pressure is via a $\text{NaSi}(\text{MgAl})_{-1}$ substitution (Gasparik, 1989). Hence, excess Si in the octahedral site of garnet appears to be an ultrahigh-pressure signature. If a majoritic garnet reequilibrates in the lithosphere, the supersilicic component exsolves as pyroxene (e.g., Haggerty and Sautter, 1990; van Roermund and Drury, 1998). Thus, for mantle-derived garnets in which pyroxene (i.e., the majoritic component) alone has exsolved, the experimental evidence supports an ultrahigh-pressure

origin for the original garnet because the exsolved pyroxene contains Si that was formerly in octahedral coordination in the garnet (e.g., Gasparik, 1989).

There are other elemental substitutions (Ti, P, Na without excess Si) which may also be indicative of very high pressures (e.g., Ringwood and Lovering, 1970; Sobolev and Lavrent'ev, 1971; Bishop et al., 1976; Reid et al., 1976). Na-, Ti- and P- rich garnets are evidenced by relatively high abundances of these elements in homogeneous garnets or exsolved phases such as rutile, ilmenite, clinopyroxene, and apatite which contain appreciable amounts of these constituents (Fig. 1). For example, garnets equilibrated (both in nature and in experiments) at moderate to high pressures and temperatures are known to contain up to 2.5 wt% TiO_2 , 1.6 wt% Na_2O and 0.6 wt% P_2O_5 (e.g., Sobolev & Lavrentyev, 1971; Thompson, 1975; Reid et al., 1976; Bishop et al., 1976, 1978; Moore and Gurney, 1985; Haggerty et al., 1994; Ono, 1998; Mproskos and Kostopoulos, 2001). Other garnets have exsolved rutile, Na-bearing diopside and apatite indicating that the initial garnet had appreciable amounts of Ti, Na and/or P (e.g., Haggerty, 1996; Roden et al., 1999; Ye et al., 2000; Mproskos and Kostopoulos, 2001). Due to the exsolution of Si-poor rutile or apatite, the original garnets may not have been supersilicic.

Some garnets from high-Mg garnet pyroxenite xenoliths from Mir have triangular arrays of mineral inclusions consistent with an origin by exsolution (Fig. 1). The inclusions are small (~100 micrometers), controlled by the garnet structure, and concentrated in the centers of grains. We have identified 6 high Mg garnet clinopyroxenites and websterites that contain garnet that is rich in these inclusions. The mineral assemblages contained in the garnets are variable, and consist of: diopside + rutile, diopside alone, rutile + high Mg (13-17 wt.% MgO) ilmenite and high Mg ilmenite alone (Table 1). Somewhat similar assemblages have been reported in garnets from the Jagersfontein kimberlite and in garnets of Norwegian orogenic peridotites (Haggerty and Sautter, 1990; Van Roermund et al., 2000). Exsolution of diopside by itself indicates that the garnet was supersilicic and formed at very high pressures. The other exsolved mineral assemblages do not require an originally supersilicic garnet, and experimental evidence is lacking to

constrain the pressure and temperature of origin. Exsolution of diopside and rutile could result from coupled substitution of Na and Ti in garnet and exsolution of ilmenite may reflect an Al- and Si- poor garnet component. These substitutions may represent high temperature rather than high pressure substitutions if the pyroxenites formed as cumulates

from an alkaline, magnesian magma intrusive into the lithosphere. Thermobarometric calculations (Brey and Kohler, 1990) for two of these pyroxenites show that they last equilibrated in the lithosphere at relatively low temperatures (750-800°C) and moderate pressures (20-30 kb, Roden et al., 1999).

Table 1: Representative Mineral Analyses

Sample	TM-90	TM-126	TM-149	TM-90	TM-126	TM-149
Mineral	discrete	discrete	discrete	host	host	host
	enstatite*	enstatite	enstatite*	garnet*	garnet	garnet*
SiO ₂	58.40	58.14	57.23	42.87	41.76	42.17
TiO ₂	0.07	0.03	0.05	0.09	0.07	0.17
Al ₂ O ₃	0.62	0.54	1.09	23.52	23.05	22.72
MgO	37.31	36.20	36.07	21.67	21.22	20.43
FeO	3.71	4.80	4.48	6.77	7.35	8.08
CaO	0.17	0.20	0.11	4.43	5.06	4.4
MnO	0.06	0.16	0.06	0.18	0.39	0.26
Na ₂ O	nd	0.05	nd	na		na
Cr ₂ O ₃	0.09	0.19	0.2	0.91	1.77	1.19
Total	100.43	100.31	99.29	100.44	100.67	99.42
Mg#	94.7	93.1	93.5			
mol% pyr				75.4	72.7	72.3

Sample	TM-90	TM-90	TM-149	TM-149	TM-126
Mineral	diopside	rutile	rutile	ilmenite	diopside
	inclusion	inclusion	inclusion	inclusion	inclusion
SiO ₂	54.76	0.07	0.18	0.11	55.09
TiO ₂	0.21	97.07	97.98	58.61	nd
Al ₂ O ₃	2.51	0.2	0.07	0.17	1.89
MgO	17.14	nd	0.02	16.78	17.18
FeO	1.09	0.49	0.61	23.58	1.58
CaO	22.15	0.19	0.22	0.18	22.93
MnO	nd	nd	0.01	0.23	0.14
Na ₂ O	1.68	na	na	na	1.36
Cr ₂ O ₃	0.64	1.23	0.93	0.66	0.81
Total	100.18	99.25	100.02	100.31	100.97
Mg#	96.6				95.1

Notes: TM-90 garnets have rutile & diopside inclusions; TM-149 garnets have rutile and ilmenite inclusions, TM-126 garnets have diopside inclusions.

*from Roden et al., 1999

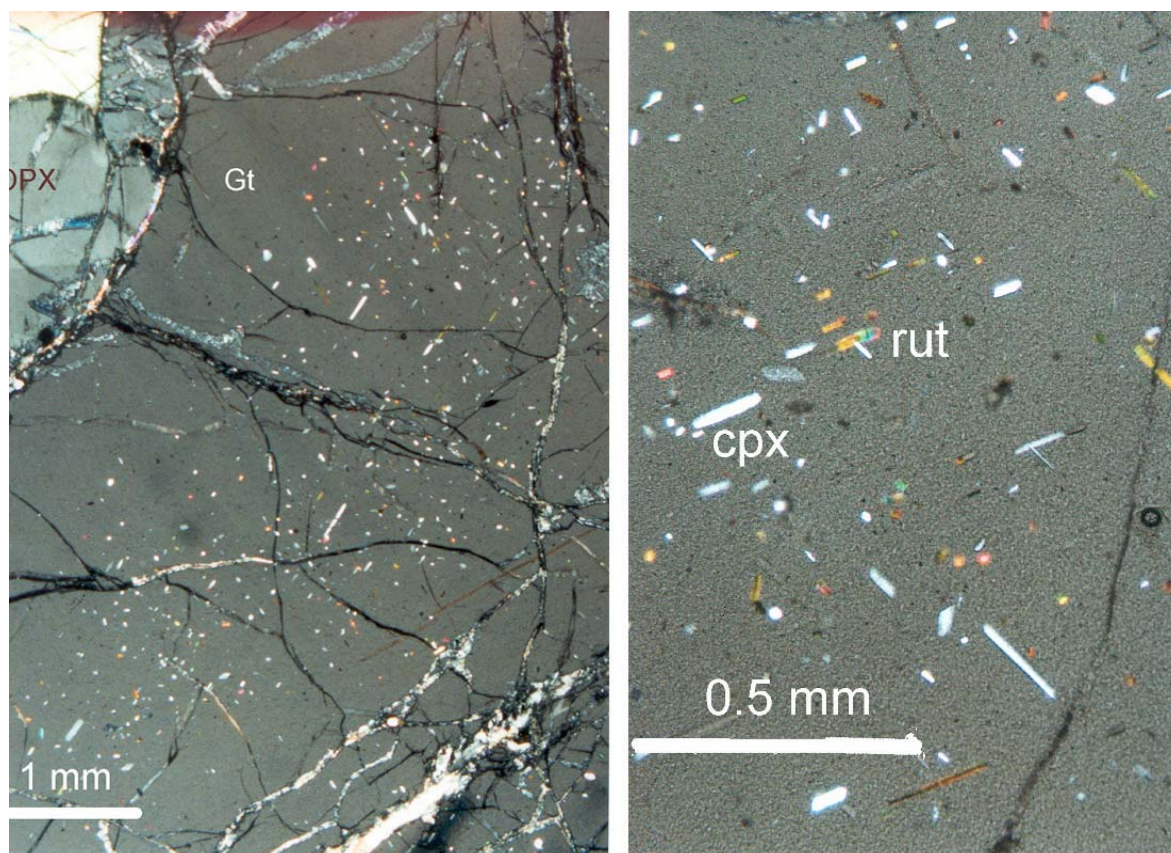


Figure 1. Diopside and rutile inclusions in garnet, garnetite TM-90.

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