XENOLITHS FROM THE ARNIE, MISERY AND PIGEON KIMBERLITES, EKATI MINE, NWT, CANADA.

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INTRODUCTION

Eighty-five peridotitic and pyroxenitic xenoliths from the **PETROGRAPHY** Arnie, Pigeon and Misery kimberlites in the Lac de Gras region, Northwest Territories, Canada have been studied. Rock types were determined using mineral associations The three kimberlites are situated within a 40 km radius of within the xenolith in conjunction with mineral compositions one another on the Ekati property (figure 1), and all are (CaO and Cr₂O₃ in garnet; Gurney, 1984). diamond-bearing.

kimberlites made for an interesting major and trace element garnet-lherzolites, whereas the pyroxenitic suite consisted of study (Doyle, 2002). Xenolith geotherms were computed using both major- and trace element geothermobarometry.

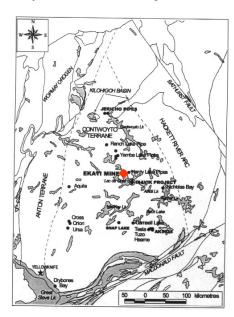


Figure 1 The three kimberlites are situated within a 40 km radius of one another on the Ekati property, NWT, Canada. (Base map: Carlson et al., 1999)

ANALYTICAL TECHNIQUES

Major and trace element analysis followed a petrographic study of the xenoliths. Major element compositions of individual minerals were determined using a wavelength dispersive electron microprobe, and trace element abundances were determined using laser ablation ICP-MS. Pressures and temperatures of equilibration were then determined using garnet-olivine, garnet-orthopyroxene and trace element geothermobarometers (T_{Ni}, P_{Cr}) .

RESULTS AND CONCLUSIONS

containing garnet were preferentially selected geothermobarometric and mineral compositional studies. The high proportion of garnet-harzburgites entrained in these The peridotite suite consisted of both garnet-harzburgites and garnet-clinopyroxenites and garnet-websterites. Harzburgites were regarded as those xenoliths that contained no clinopyroxene (Dawson, 1981).

> Peridotite xenoliths were dominant at all three kimberlite localities. Spinel exsolution from ortho- and clinopyroxene in both peridotitic and pyroxenitic xenoliths from the Misery and Arnie kimberlites is indicative of partial re-equilibration of these xenoliths from a region of high pressure and temperature to lower pressure and temperature (Harte and Gurney, 1975). Such dynamic processes and small scale heterogeneity suggest a complex history for the lithosphere beneath the Slave province.

MAJOR AND TRACE ELEMENT MINERAL CHEMISTRY

Pyroxenitic minerals are more Fe-rich and Cr₂O₃-poor than the peridotitic minerals (Doyle, 2002). The harzburgitic xenoliths were divided into two sub-groups based on CaO and Cr₂O₃ content in garnet (Gurney, 1984), namely a high-Ca group and a low-Ca group. Major and trace element compositions of selected peridotitic and pyroxenitic xenoliths are detailed in tables 1 and 2, and depicted in figures 2, 3 and 4. The major element composition of the high-Ca harzburgitic garnets are similar to the lherzolitic garnets (figure 3), whereas the trace element characteristics of the high-Ca harzburgitic garnets are more alike the low-Ca harzburgites (figure 4).

The lherzolitic and pyroxenitic garnets have chondritenormalised LREE-depleted patterns (figure 5), whereas the harzburgitic garnets have sinusoidal and MREE-enriched patterns in addition to the LREE-depleted pattern. Many authors have speculated about the origin of the MREEenriched garnet pattern (Hoal et al., 1994; Shimizu and Sobelev, 1995; Griffin et al., 1999(a)), but not the sinusoidal Doyle (2002) tentatively speculates that the sinusoidal pattern of the pyrope may be retained from the

exsolution of pyrope and orthopyroxene from a pre-existing garnets (both low-Ca and high-Ca) were more enriched in majorite component. normalized garnet pattern would however be produced if figures 4 and 5), whereas the lherzolitic garnets were more pyrope and clinopyroxene were the exsolution products.

The trace elements Nb, Y, Zr and Ti discriminated well than the pyroxenitic garnets (figure 6). between the different xenoliths: In general the harzburgitic

The LREE-depleted chondrite Nb than the lherzolitic and pyroxenitic garnets (table 2, enriched in Y and Ti than the harzburgitic and pyroxenitic garnets. The lherzolitic garnets have higher Zr abundances

Table 1: Major element compositions of garnet of selected xenoliths.

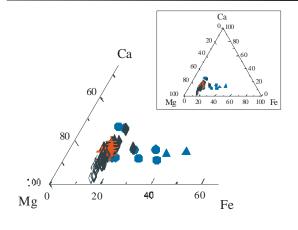
Suite:	Low-Ca	Low-Ca	Low-Ca	Low-Ca	High-CA	High-Ca	Lhz	Lhz	Clino	Web
SampleNo.	ARN007	ARN008	ARN025	ARN026	ARN003	ARN006	MIS409	PGN337	ARN028	MIS438
SiO ₂	41.36	42.18	41.59	42.28	40.93	41.20	41.68	42.28	40.59	38.91
TiO ₂	<0.04*	< 0.04	< 0.04	0.04	< 0.04	< 0.04	< 0.04	0.62	0.11	0.08
Al ₂ O ₃	18.20	20.11	18.97	20.96	17.94	19.70	21.24	20.77	19.67	22.26
Cr ₂ O ₃	7.12	4.91	7.26	4.70	8.02	5.50	3.64	3.39	4.17	0.44
FeO	6.99	7.78	7.07	7.36	7.06	8.22	8.16	8.50	12.66	22.03
MnO	0.42	0.43	0.42	0.40	0.48	0.58	0.41	0.35	0.48	0.51
MgO	22.63	22.47	20.95	20.49	19.19	18.92	19.36	19.47	16.38	10.25
CaO	2.96	1.35	3.49	3.96	6.60	5.92	5.17	4.92	6.09	5.66
Na ₂ O	0.04	0.03	0.04	0.02	0.04	0.04	0.02	ND	0.02	0.02
Total	99.73	99.27	99.79	100.21	100.27	100.09	99.68	100.48	100.18	100.14
Mg#	85.23	83.74	84.09	83.22	82.90	80.41	80.89	80.33	69.76	45.35
# Anal	7	9	11	5	4	5	5	4	8	3

Key: <0.04 = <Lower Limit of Detection. Low-Ca = Low-Ca harzburgite, High-Ca = High-Ca harzburgite, Lhz = Lherzolite, Clino = Clinopyroxenite, Web = Websterite

Table 2: Trace element abundances in garnet of selected xenoliths, NWT

Suite	Low-Ca	Low-Ca	Low-Ca	High-Ca	High-Ca	Lhz	Lhz	Pyrox
Sample No	ARN007	ARN025	PGN314	ARN003	ARN004	MIS409	PGN337	ARN028
Ti	56	<25.62*	371	125	38	567	2850	537
Sr	1.2	1.8	0.57	1.8	1.8	0.22	0.54	< 0.37
Y	0.65	< 0.71	2.1	35	1.2	8.6	12	9.6
Zr	32	53	29	107	3.3	4.7	27	1.9
Nb	0.33	0.50	< 0.22	1.3	0.090	0.34	0.44	0.77
La	0.13	0.92	0.44	0.25	0.67	0.05	< 0.43	< 0.43
Ce	1.3	2.3	0.44	2.0	5.4	0.15	0.16	0.27
Nd	5.2	9.4	1.4	7.7	6.2	0.27	0.75	0.54
Sm	3.4	5.1	0.50	5.5	0.92	0.36	0.54	< 0.26
Eu	0.94	1.3	0.58	2.4	0.25	0.24	< 0.54	< 0.54
Gd	2.2	3.6	0.85	10.2	0.49	1.2	1.3	< 0.41
Dy	0.39	0.68	0.79	8.8	0.16	1.8	2.1	1.7
Er	< 0.054	0.66	0.57	3.2	0.15	1.2	1.4	1.3
Yb	0.16	0.30	0.56	2.2	0.26	1.3	1.6	1.3
Garnet REE pattern	MREE- enriched	Sinusoidal	LREE- depleted	MREE- enriched	Sinusoidal	LREE- depleted	LREE- depleted	LREE - depleted

Key: <25.26 = <Lower Limit of Detection. Suite definitions as for Table 1 (see above).



Peridotite Suite

Low-Ca Harzburgite
High-Ca Harzburgite
High-Ca Harzburgite
Lherzotite
Pyroxenite Suite
Clinopyroxenite
Websterite

Cao (wt%)

Figure 2 Fe-Mg-Ca of garnet from mantle xenoliths from three kimberlite localities, NWT (symbols as for figure 3).

Figure 3 CaO vs $Cr_2.O_3$ in garnet from mantle xenoliths from the Arnie, Pigeon and Misery kimberlites, NWT.

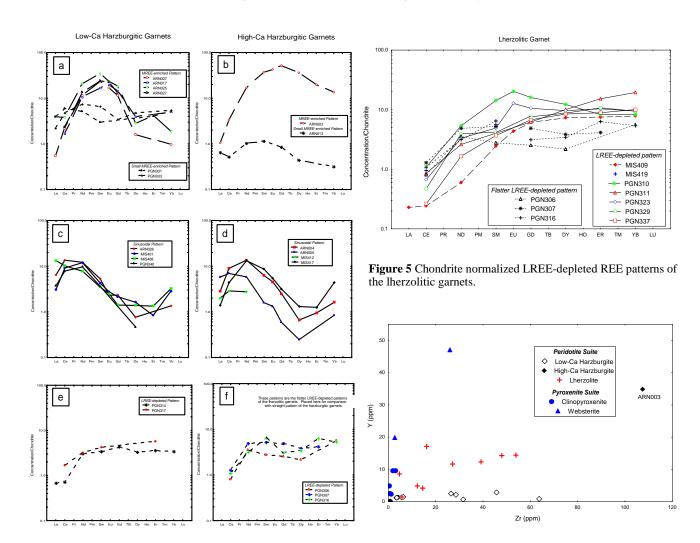


Figure 4 Chondrite normalized REE-patterns of the low-ca and high-Ca harzburgitic garnets. The low-Ca and high-Ca harzburgitic garnets have similar REE characteristics.

Figure 6 Y vs Zr in garnet. The lherzolitic garnets are more enriched in Y than the harzburgitic garnets, and more enriched in Zr than the pyroxenitic garnets.

Major and Trace Element Geothermobarometry

peridotitic xenoliths define a 40 mW.m⁻² geotherm (Pollack and Chapman, 1977). Major element geothermobarometry field (Kennedy and Kennedy, 1976). indicates that although there are geochemical similarities between the low-Ca garnets in harzburgites and peridotitic Pioneering studies of the Slave craton have recognised garnets included in diamond, not all low-Ca harzburgites layering within the underlying lithospheric mantle (Griffin et have pressures and temperatures of equilibration in the al., 1999 (b); Pearson et al., 1999; Kopylova and Russell, diamond stability field (table 3, figure 7). Low-Ca 2000). This study suggests that the harzburgites and harzburgitic xenoliths with <6 wt% Cr₂O₃ in garnet (e.g. lherzolites are inter-mixed rather than separated into two ARN008, ARN026) have shallow pressures and distinct layers (figure 8). temperatures of equilibration. In contrast, the low-Ca and

high-Ca harzburgitic xenoliths with >8 wt% Cr₂O₃ (e.g. ARN002, PGN312) have pressures and temperatures of Major and trace element geothermobarometry reveal that the equilibration within the diamond stability field. Xenoliths with 6-8 wt% Cr₂O₃ straddle the graphite-diamond stability

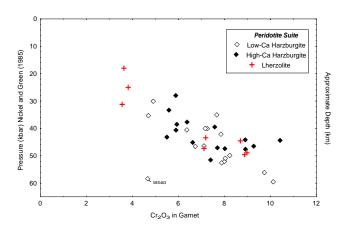


Figure 7 Pressure (Nickel and Green, 1985) vs Cr₂O₃ in garnet. There is a direct relationship between Cr₂O₃ content in garnet and pressure, as described in the text.

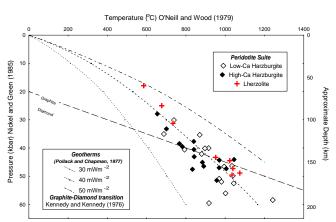


Figure 8 Pressure vs temperature. The lherzolites and harzburgites from the Pigeon, Arnie and Misery kimberlite localities appear to be interlayered, lying along

Table 3: Temperature, Pressure and Cr₂O₃ content in garnet of selected xenoliths, NWT

Kimberlite	Sample No.	O'Neill & Wood	Nickel & Green	Cr ₂ O ₃ content	Rock Suite
Locality		gt-ol (°C)	gt-opx (Kbar)	in Garnet	
Misery	MIS426	735	31.2	3.56	Lherzolite
Arnie	ARN026	871	35.3	4.70	Low-Ca Harzburgite
Arnie	ARN008	738	30.1	4.91	Low-Ca Harzburgite
Misery	MIS430	768	38.5	5.93	High-Ca Harzburgite
Arnie	ARN007	1035	46.4	7.12	Low-Ca Harzburgite
Pigeon	PGN339	1039	47.3	7.12	Lherzolite
Arnie	ARN003	1006	47.4	8.02	High-Ca Harzburgite
Arnie	ARN002	969	50.9	8.04	Low-Ca Harzburgite
Pigeon	PGN314	1039	49.9	8.24	Low-Ca Harzburgite
Pigeon	PGN316	1024	44.5	8.69	Lherzolite
Pigeon	PGN312	1045	44.1	8.91	High-Ca Harzburgite
Misery	MIS408	917	59.5	10.13	Low-Ca Harzburgite

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